

COMPARISON OF EUROPEAN, US & JAPANESE PHARMACOPOEIA MONOGRAPHS FOR MEDICINAL GASES

Doc 152/23

Revision of Doc 152/18

EUROPEAN INDUSTRIAL GASES ASSOCIATION AISBL 

AVENUE DE L'ASTRONOMIE 30 • B-1210 BRUSSELS
Tel: +32 2 217 70 98
E-mail: info@eiga.eu • Internet: www.eiga.eu



COMPARISON OF EUROPEAN, US & JAPANESE PHARMACOPOEIA MONOGRAPHS FOR MEDICINAL GASES

Prepared by WG-7 Medical and Breathing Gases

Disclaimer

All technical publications of EIGA or under EIGA's name, including Codes of practice, Safety procedures and any other technical information contained in such publications were obtained from sources believed to be reliable and are based on technical information and experience currently available from members of EIGA and others at the date of their issuance.

While EIGA recommends reference to or use of its publications by its members, such reference to or use of EIGA's publications by its members or third parties are purely voluntary and not binding.

Therefore, EIGA or its members make no guarantee of the results and assume no liability or responsibility in connection with the reference to or use of information or suggestions contained in EIGA's publications.

EIGA has no control whatsoever as regards, performance or non performance, misinterpretation, proper or improper use of any information or suggestions contained in EIGA's publications by any person or entity (including EIGA members) and EIGA expressly disclaims any liability in connection thereto.

EIGA's publications are subject to periodic review and users are cautioned to obtain the latest edition.



Table of Contents

1	Introduction	1
2	Scope and purpose	1
2.1	Scope	1
2.2	Purpose	2
3	Specifications and test methods	3
3.1	European Pharmacopoeia test requirements	4
4	Currency of information	5
5	European Pharmacopoeia compared to United States Pharmacopoeia	5
5.1	Medical oxygen	6
5.2	93% Oxygen	7
5.3	Nitrous oxide	8
5.4	Nitrogen	9
5.5	97% Nitrogen	10
5.6	Nitrogen, low-oxygen	11
5.7	Carbon dioxide	12
5.8	Air, medicinal	13
5.9	Air, synthetic medicinal	14
5.10	Helium	15
5.11	Nitric oxide	16
5.12	Argon	17
5.13	Carbon monoxide	18
5.14	Carbon monoxide intermix (5 per cent) in nitrogen	19
5.15	Methane	20
5.16	Methane intermix (2% per cent) in nitrogen	21
5.17	Acetylene intermix (1 per cent) in nitrogen	22
6	Japanese Pharmacopoeia (18th Edition)	23
6.1	Oxygen	23
6.2	Nitrous oxide	24
6.3	Carbon dioxide	25
6.4	Nitrogen	26

Amendments from 152/18

Section	Change
3.1	Clarification relating to intended use of the sections of the monographs
4	Update of European and Japanese Pharmacopoeias
5	General update to reflect the use of the various test methods and update of the monographs to the latest version
6	Analytical method for Oxygen assay modified in Japanese Pharmacopoeia and inclusion of the type of detector for GC

1 Introduction

There are three prime regional Pharmacopoeia organisations that are responsible for the preparation and publication of Pharmacopoeia monographs, covering the commonly used substances involved in the manufacture and supply of medicinal products.

The three organisations are:

- European Directorate for the Quality of Medicines (EDQM), who are responsible for the European Pharmacopoeia (Ph. Eur.) monographs
- United States Pharmacopoeia Convention, who are responsible for the US Pharmacopoeia (USP)
- Ministry of Health and Welfare, who are responsible for the Japanese Pharmacopoeia (JP)

The three Pharmacopoeias have monographs for a number of medical / medicinal gases. These gases can be used either as active ingredients in medicinal products or excipients, used in the manufacture of medical gas mixtures, administered to patients. Alternatively, they can be used as pharmaceutical gases, used in the manufacture, storage or distribution of all medicinal products.

The purpose of these monographs is to specify for each gas:

- Minimum assay / purity for the product that is suitable for medicinal use;
- Maximum level of defined impurities, that could have an adverse effect on the patient; and
- Appropriate test methods for determining quality of the product.

This publication provides a comparison between the specifications and the test methods defined in each of the regional pharmacopoeia compendiums.

2 Scope and purpose

2.1 Scope

This publication covers the pharmacopoeia monographs for medicinal and pharmaceutical gases published by the:

- European Pharmacopoeia;
- United States Pharmacopoeia; and
- Japanese Pharmacopoeia.

It includes the monographs for gases used in the manufacture and supply of medicinal products including:

- Medicinal gases, that are used as active ingredients in medical gases and gas mixtures supplied for patient use;
- Excipient gases, that are added to gas mixtures but have no pharmacological effects; and
- Pharmaceutical gases, that are specified in the manufacture, storage and distribution of medicinal products.

The comparison tables provide a comparison between the European and the United States Pharmacopoeia monographs for all of the specified gases.

A separate table is included to detail the monographs published by the Japanese Pharmacopoeia, where the monographs do not specify acceptance limits and only provide test criteria for compliance.

2.2 Purpose

To provide a cross reference between the three sets of published monographs to enable a comparison of the requirements for each method. This is intended to demonstrate compliance between monographs but should not be used as a detailed method of carrying out the relevant tests.

Where the testing to a specific monograph is required, the user should refer to the original document (and all supporting documents within the relevant pharmacopoeia) to ensure that the tests are carried out correctly.

3 Specifications and test methods

The most commonly used medicinal gases have been included in the Pharmacopoeia for many years but recently a number of new medical gases have been added.

The following table gives a reference for the different gases that have been covered by published monographs in the three Pharmacopoeias:

Gases	Monograph reference		
	European Pharmacopoeia	US Pharmacopoeia	Japanese Pharmacopoeia
Medical oxygen	0417	No reference no.	No reference no.
Oxygen 93%	2455	No reference no.	NS
Nitrous oxide	0416	No reference no.	No reference no.
Nitrogen	1247	No reference no.+	No reference no.
Nitrogen 97%	NS	No reference no.	NS
Nitrogen, low-oxygen	1685	No reference no.	NS
Carbon dioxide	0375	No reference no.	No reference no.
Air, medicinal	1238	No reference no.	NS
Air, synthetic medicinal	1684	See medicinal air	NS
Helium	2155	NS	NS
Nitric oxide	1550	NS	NS
Argon	2407	NS	NS
Carbon monoxide	2408	NS	NS
Carbon monoxide intermix (5 per cent) in nitrogen	2904	NS	NS
Methane	2413	NS	NS
Methane intermix (2 per cent) in nitrogen	2905	NS	NS
Acetylene intermix (1 per cent) in nitrogen	2903	NS	NS

+ Nitrogen is covered in the National Formulary

NS: Not specified

Each monograph defines the specification of the medicinal gas, including the:

- Assay of the product;
- Maximum allowable impurity levels for those contaminants specified in the product;
- Approved analytical method for identifying the gas;
- Approved analytical method for determining the assay, and

- Approved analytical test method for determining each contaminant specified within the monograph.

The validated analytical methods described in the monographs are the official test methods upon which the specifications in the relevant Pharmacopoeia are based.

3.1 European Pharmacopoeia test requirements

For the European Pharmacopoeia, the test methods are verified against the protocols set out in the International Council for Harmonisation of Technical Requirements for Pharmaceuticals for Human Use, (ICH) guidelines for accuracy and precision, linearity and range and specificity. The results also need to conform to the requirements of repeatability and peak symmetry.

In addition to the specific medicinal gas monographs there are several general notices that apply to all monographs.

The following general test methods are particularly applicable to the analysis of medical gases:

Test reference	Test method
2.1.6	Gas detector tubes
2.2.24	Absorption spectrophotometry, infrared
2.2.28	Gas chromatography
2.2.46	Chromatographic separation techniques
2.5.24	Carbon dioxide in gases
2.5.25	Carbon monoxide in gases
2.5.26	Nitrogen monoxide and nitrogen dioxide in gases
2.5.27	Oxygen in gases
2.5.28	Water in gases
2.5.35	Nitrous oxide in gases

Alternative methods of analysis can be used for testing medical gases, after agreement with the competent authority, provided that the test methods have been validated in line with the ICH protocols to demonstrate that they are equivalent to the specified methods.

In the medicinal gases monographs where there is a production and tests section it was originally intended that the test section can be used by the end user e.g. the hospital, whereas the production section is intended for the manufacturer. However, this structure is not present in all the medicinal gases monographs. Therefore, in this document to avoid any confusion we named the sections "Tests intended for the manufacturer" and "Tests intended for the end user".

Where the hospital is the manufacturer, for example where they are producing medicinal gases on site (e.g. Oxygen 93 %), the "Tests intended for the manufacturer" shall be applied.

The United States and Japanese Pharmacopoeia monographs only specify one method for the analysis of the medical gas.

The Japanese Pharmacopoeia monographs only detail the test methods and do not give the values of the specification limits for the impurities in percentage terms or parts per million. The approved test methods include either gas chromatography, detector tubes or wet chemistry as the approved methods.

In all cases the test methods specified in the monographs should have been validated. For the European Pharmacopoeia, this work is normally undertaken by one of the national representatives on the relevant Pharmacopoeia committee.

4 Currency of information

The versions of the relevant Pharmacopoeias used to provide the information for the comparison tables are:

- European Pharmacopoeia: 11.0 (2023)
- United States Pharmacopoeia: USP40, NF35 (2017)
- Japanese Pharmacopoeia: 18 (2021)

As and when there are relevant changes to any of these monographs this publication shall be updated. However, where it is important that the latest information is available, the original Pharmacopoeia document should be referenced to ensure that there have been no revisions to the individual monograph.

5 European Pharmacopoeia compared to United States Pharmacopoeia

The following tables provide a comparison between the European and US Pharmacopoeia monographs for each of the specified medicinal or pharmaceutical gas.

5.1 Medical oxygen

Oxygen			
Monograph		Ph. Eur.	USP
Name		Oxygen	Oxygen
Reference		01/2010:0417	Not specified
Chemical formula		O ₂	O ₂
Definition		Oxygen contains not less than 99.5% V/V of oxygen. It is produced by a purification process followed by a cryodistillation of the ambient air.	Oxygen contains not less than 99.0% V/V of oxygen. Note: Oxygen produced by the air-liquefaction is exempt from the requirements of carbon monoxide and carbon dioxide testing.
Identification		Complies with the assay	Complies with the assay
Tests intended for the manufacturer			
Assay	Specification	≥ 99.5% V/V oxygen	≥ 99.0 % V/V oxygen
	Analytical method	Paramagnetic analyser	Paramagnetic analyser
Impurities			
CO	Limit	≤ 5 ppm V/V	≤ 0.001% V/V
	Analytical method	Infrared analyser	Detector tube
CO₂	Limit	≤ 300 ppm V/V	≤ 0.03% V/V
	Analytical method	Infrared analyser	Detector tube
H₂O	Limit	≤ 67 ppm V/V	Not specified
	Analytical method	Electrolytic hygrometer	
Tests intended for the end user			
CO	Limit	≤ 5 ppm V/V	Not specified
	Analytical method	Detector tube	
CO₂	Limit	≤ 300 ppm V/V	
	Analytical method	Detector tube	
H₂O	Limit	≤ 67 ppm V/V	
	Analytical method	Detector tube	

5.2 93% Oxygen

Oxygen 93			
Monograph		Ph. Eur.	USP
Name		Oxygen (93 per cent)	Oxygen 93 Percent
Reference		04/2011:2455	Not Specified
Chemical formula		O ₂	O ₂
Definition		Oxygen 93% contains between 90.0%V/V and 96% V/V of oxygen. Remainder mainly consists of argon and nitrogen. Monograph applies to oxygen 93% produced in single-stage concentrators by absorption purification of ambient air using zeolites. It does not apply to gas produced using individual concentrators for domiciliary use.	Oxygen 93 is oxygen produced from air by molecular sieve process. Contains not less than 90.0 % V/V and not more than 96 % oxygen V/V, the remainder consists of mostly argon and nitrogen
Identification		Complies with the assay	Complies with the assay
Tests intended for the manufacturer			
Assay	Specification	90.0% ≤ O ₂ ≤ 96.0% V/V oxygen	90.0% ≤ O ₂ ≤ 96.0% V/V oxygen
	Analytical method	Paramagnetic analyser	Paramagnetic analyser
Impurities			
CO	Limit	≤ 5 ppm V/V	≤ 0.001 % V/V
	Analytical method	Infrared analyser	Detector tube
CO₂	Limit	≤ 300 ppm V/V	≤ 0.03 % V/V
	Analytical method	Infrared analyser	Detector tube
H₂O	Limit	≤ 67 ppm V/V	Not specified
	Analytical method	Electrolytic hygrometer	
Odour	Limit		No odour
	Analytical method		Organoleptic
NO/ NO₂	Limit	≤ 2 ppm V/V in total	Not specified
	Analytical method	Chemiluminescence analyser	
SO₂	Limit	≤ 1 ppm V/V	
	Analytical method	UV Fluorescence analyser	
Oil	Limit	≤ 0.1 mg/m ³	
	Analytical method	Detector tube	
Tests intended for the end user			
CO	Limit	≤ 5 ppm V/V	Not specified
	Analytical method	Detector tube	
CO₂	Limit	≤ 300 ppm V/V	
	Analytical method	Detector tube	
H₂O	Limit	≤ 67 ppm V/V	
	Analytical method	Detector tube	
NO/ NO₂	Limit	≤ 2 ppm V/V	
	Analytical method	Detector tube	
SO₂	Limit	≤ 1 ppm V/V	
	Analytical method	Detector tube	
Oil	Limit	≤ 0.1 mg/m ³	
	Analytical Method	Detector tube	

5.3 Nitrous oxide

Nitrous oxide			
Monograph		Ph. Eur.	USP
Name		Nitrous oxide	Nitrous oxide
Reference		01/2008:0416	Not specified
Chemical Formula		N ₂ O	N ₂ O
Definition		Contains not less than 98.0% V/V of nitrous oxide in the gaseous phase, when sampled at 15°C. Nitrous oxide is produced from ammonium nitrate by thermic decomposition.	Nitrous Oxide contains not less than 99.0% V/V of nitrous oxide
Identification		Complies with the assay. or - Place a glowing splinter of wood in the substance to be examined. The splinter bursts into flame. or - Introduce the substance to be examined into alkaline pyrogallol solution R. A brown colour does not develop.	Comparison of pressure between nitrous oxide container and certified standard. Distinction from carbon dioxide detector tube. Distinction from oxygen (alkaline pyrogallol solution)
Tests intended for the manufacturer			
Assay	Assay	≥ 98.0% V/V nitrous oxide Measured in gas phase at 15°C	≤ 1.0% air indicating ≥ 99.0% V/V of nitrous oxide
	Analytical method	Infrared analyser	Gas chromatography
Impurities			
CO	Limit	≤ 5 ppm V/V	≤ 0.001% V/V
	Analytical method	Gas chromatography with flame ionisation with methaniser detector	Detector tube
CO ₂	Limit	≤ 300 ppm V/V	≤ 0.03% V/V
	Analytical method	Gas chromatography with thermal conductivity detector	Detector tube
NO/ NO ₂	Limit	≤ 2 ppm V/V in total in the gaseous and liquid phases	Nitric oxide ≤ 1ppm, nitrogen dioxide ≤ 1ppm
	Analytical method	Chemiluminescence analyser	Detector tube
H ₂ O	Limit	≤ 67 ppm V/V	≤ 150 mg/m ³
	Analytical method	Electrolytic hygrometer	Detector tube
NH ₃	Limit	Not specified	≤ 0.0025 % V/V
	Analytical method		Detector tube
Halo- gens	Limit	Not specified	≤ 1ppm
	Analytical method		Detector tube
Tests intended for the end user			
CO	Limit	≤ 5 ppm V/V	Not specified
	Analytical method	Detector tube	
CO ₂	Limit	≤ 300 ppm V/V	
	Analytical method	Detector tube	
NO/ NO ₂	Limit	≤ 2 ppm V/V	
	Analytical method	Detector tube	
H ₂ O	Limit	≤ 67 ppm V/V	
	Analytical method	Detector tube	

5.4 Nitrogen

Nitrogen			
Monograph		Ph. Eur.	USP
Name		Nitrogen	Nitrogen
Reference		01/2023:1247	7727-37-9
Chemical formula		N ₂	N ₂
Definition		Nitrogen contains not less than 99.5% V/V of nitrogen.	Nitrogen contains not less than 99.0%, by volume of nitrogen
Identification		Comparison of peak retention times of sample and reference gas obtained with gas chromatography or - Place a glowing splinter of wood in the substance to be examined. The splinter is extinguished. or - Test with magnesium turnings	Extinguishing of burning wood splinter in a nitrogen test tube.
Tests intended for the manufacturer			
Assay	Assay	≥ 99.5% V/V nitrogen	≤ 1.0% oxygen indicates ≥ 99.0% V/V of nitrogen
	Analytical method	Gas chromatography with thermal conductivity detector	Gas chromatography
Impurities			
CO	Limit	≤ 5 ppm V/V	≤ 0.001 % V/V
	Analytical method	Infrared analyser	Detector tube
CO ₂	Limit	≤ 300 ppm V/V	Not specified
	Analytical method	Infrared analyser	
O ₂	Limit	≤ 50 ppm V/V	≤ 1.0 %
	Analytical method	Oxygen analyser with electrochemical cell	Determined in assay
H ₂ O	Limit	≤ 67 ppm V/V	Not specified
	Analytical method	Electrolytic hygrometer	
Odour	Limit	Not specified	No odour
	Analytical method		Organoleptic
Tests intended for the end user			
CO	Limit	≤ 5 ppm V/V	Not specified
	Analytical method	Detector tube	
CO ₂	Limit	≤ 300 ppm V/V	
	Analytical method	Detector tube	
H ₂ O	Limit	≤ 67 ppm V/V	
	Analytical method	Detector tube	

5.5 97% Nitrogen

Nitrogen 97%			
Monograph		Ph. Eur. No equivalent European Pharmacopoeia monograph	USP
Name			Nitrogen 97 Percent
Reference			Not Specified
Chemical formula			N ₂
Definition			Nitrogen 97 % is nitrogen produced from air by physical separation. Contains not less than 97.0% nitrogen V/V
Identification			Extinguishing of burning wood splinter in a nitrogen test tube.
Tests intended for the manufacturer			
Assay	Assay		≤ 3.0% oxygen indicates ≥ 97.0% V/V of nitrogen
	Analytical method		Gas chromatography
Impurities			
CO	Limit		≤ 0.001 % V/V
	Analytical method		Detector tube
CO₂	Limit		≤ 0.03 % V/V
	Analytical method		Detector tube
SO₂	Limit		≤ 5 ppm V/V
	Analytical method		Detector tube
NO/ NO₂	Limit		≤ 2.5 ppm V/V
	Analytical method		Detector tube
O₂	Limit		≤ 3.0 % V/V
	Analytical method		Gas chromatography (determined in the assay)
Odour	Limit		No odour
	Analytical method		Organoleptic
Tests intended for the end user			
	Limit		Not specified
	Analytical method		

5.6 Nitrogen, low-oxygen

Nitrogen, low-oxygen			
Monograph	Ph. Eur.		USP No equivalent US Pharmacopoeia monograph
Name	Nitrogen low oxygen		
Reference	01/2008:1685 corrected 11.0		
Chemical formula	N ₂		
Definition	<p>This monograph applies to nitrogen which is used for inerting finished medicinal products which are particularly sensitive to degradation by oxygen.</p> <p>Does not necessarily apply to nitrogen used in earlier production steps of pharmaceutical manufacturing.</p>		
Identification	<p>Examine the chromatograms obtained in the test for impurities.</p> <p>or - Flame of burning wood splinter in a nitrogen test tube/ test with magnesium turnings.</p>		
Tests intended for the manufacturer			
Assay	Assay	≥ 99.5% V/V nitrogen	
	Analytical method	Gas chromatography with thermal conductivity detector	
Impurities			
O₂	Limit	≤ 5 ppm V/V	
	Analytical method	Oxygen analyser with electrochemical cell	
Total impurities	Limit	≤ 0.5% V/V	
	Analytical method	Gas chromatography with thermal conductivity detector	
Tests intended for the end user			
	Limit	No test section specified	

5.7 Carbon dioxide

Carbon dioxide			
Monograph		Ph. Eur.	USP
Name		Carbon dioxide	Carbon dioxide
Reference		01/2008:0375	Not specified
Chemical Formula		CO ₂	CO ₂
Definition		Carbon dioxide contains not less than 99.5% V/V carbon dioxide in gaseous phase.	Carbon dioxide contains not less than 99.0%, by volume of carbon dioxide
Identification		Infrared absorption spectrophotometry <i>or</i> - glowing wood splinter extinguished <i>or</i> - test with barium hydroxide solution R and dilute acetic acid R	Carbon dioxide detector tube
Tests intended for the manufacturer			
Assay	Assay	≥ 99.5% V/V carbon dioxide	≥ 99.0% V/V of carbon dioxide
	Analytical method	Infrared analyser	Determined with volumetric gas absorption apparatus
Impurities			
CO	Limit	≤ 5 ppm V/V	≤ 0.001% V/V
	Analytical method	Gas chromatography with flame ionisation methaniser detector	Detector tube
NO/ NO ₂	Limit	≤ 2 ppm V/V in total (in gas phase)	NO ≤ 2.5 ppm (in gas phase) NO ₂ ≤ 2.5 ppm (in liquid phase)
	Analytical method	Chemiluminescence analyser	Detector tube
Total Sulfur	Limit	≤ 1 ppm V/V	Not specified
	Analytical method	UV fluorescence analyser	
H ₂ O	Limit	≤ 67 ppm V/V	≤ 150 mg/m ³
	Analytical method	Electrolytic hygrometer	Detector tube
NH ₃	Limit	Not specified	≤ 0.0025 % V/V
	Analytical method		Detector tube
H ₂ S	Limit	Not specified	≤ 1 ppm
	Analytical method		Detector tube
SO ₂	Limit	Not specified	≤ 5 ppm
	Analytical method		Detector tube
Tests intended for the end user			
CO	Limit	≤ 5 ppm V/V	Not specified
	Analytical method	Detector tube	
SO ₂	Limit	≤ 2 ppm V/V	
	Analytical method	Detector tube	
H ₂ S	Limit	≤ 1 ppm V/V	
	Analytical method	Detector tube	
NO/ NO ₂	Limit	≤ 2 ppm V/V	
	Analytical method	Detector tube	
H ₂ O	Limit	≤ 67 ppm V/V	
	Analytical method	Detector tube	

5.8 Air, medicinal

Air, medicinal			
Monograph		Ph. Eur.	USP
Name		Air, medicinal	Medical air
Reference		07/2022:1238	Not specified
Chemical formula		N/A	N/A
Definition		Compressed ambient air containing not less than 20.4 %V/V and not more than 21.4 % V/V of oxygen.	Natural or synthetic mixture consisting largely of nitrogen and oxygen, containing not less than 19.5% and not more than 23.5% V/V of oxygen.
Identification		Complies with the assay or - glowing wood splinter not extinguished or - tested by passing sample through potassium hydroxide /sodium dithionite solution.	Meets the assay acceptance criteria
Tests intended for the manufacturer			
Assay	Assay	$20.4\%V/V \leq \text{oxygen} \leq 21.4\% V/V$	$19.5\% V/V \leq \text{oxygen} \leq 23.5\% V/V$
	Analytical method	Paramagnetic analyser	Paramagnetic analyser*
Impurities			
CO	Limit	$\leq 5 \text{ ppm V/V}$	$\leq 0.001\% V/V^*$
	Analytical method	Infrared analyser	Detector tube
CO ₂	Limit	$\leq 500 \text{ ppm V/V}$	$\leq 0.05\% V/V^*$
	Analytical method	Infrared analyser	Detector tube
SO ₂	Limit	$\leq 1 \text{ ppm V/V}$	$\leq 5 \text{ ppm V/V}^*$
	Analytical method	UV fluorescence analyser	Detector tube
NO/ NO ₂	Limit	$\leq 2 \text{ ppm V/V in total}$	$\leq 2.5 \text{ ppm V/V}^*$
	Analytical method	Chemiluminescence analyser	Detector tube
Oil	Limit	$\leq 0.1 \text{ mg/m}^3$	No condensate on mirror
	Analytical method	Detector tube when oil lubricated compressor is used	Pass gas slowly over stainless steel mirror*
H ₂ O	Limit	$\leq 67 \text{ ppm V/V}$	No condensate on mirror
	Analytical method	Electrolytic hygrometer	Pass gas slowly over stainless steel mirror*
Tests intended for the end user			
CO	Limit	$\leq 5 \text{ ppm V/V}$	Not specified
	Analytical method	Detector tube	
CO ₂	Limit	$\leq 500 \text{ ppm V/V}$	
	Analytical method	Detector tube	
SO ₂	Limit	$\leq 1 \text{ ppm V/V}$	
	Analytical Method	Detector Tube	
NO/ NO ₂	Limit	$\leq 2 \text{ ppm V/V}$	
	Analytical Method	Detector Tube	
Oil	Limit	$\leq 0.1 \text{ mg/m}^3$	
	Analytical Method	Detector Tube	
H ₂ O	Limit	$\leq 67 \text{ ppm V/V}$	
	Analytical Method	Detector Tube	

* Not required for synthetic air if so labelled

5.9 Air, synthetic medicinal

Air, synthetic medicinal		
Monograph	Ph. Eur.	USP No equivalent US Pharmacopoeia monograph specified, but covered by medical air
Name	Air, synthetic medicinal	
Reference	01/2008:1684	
Chemical formula	N/A	
Definition	Gas mixture of nitrogen (Ph. Eur) and oxygen (Ph.Eur) containing between 95.0 % to 105.0 % of the nominal value which is between 21.0 % V/V to 22.5 % V/V of oxygen.	
Identification	Complies with the assay or -glowing wood splinter not extinguished or - oxygen content tested by passing sample through potassium hydroxide/sodium dithionite solution.	
Tests intended for the manufacturer		
Assay	Assay	Containing between 95.0% to 105.0% of the nominal value which is between 21.0 % V/V to 22.5 % V/V of oxygen.
	Analytical method	Paramagnetic analyser
Impurities		
H₂O	Limit	≤ 67 ppm V/V
	Analytical method	Electrolytic hygrometer
Tests intended for the end user		
H₂O	Limit	≤ 67 ppm V/V
	Analytical method	Detector tube

5.10 Helium

Helium			
Monograph		Ph. Eur.	USP
Name		Helium	Helium
Reference		01/2008:2155	Not specified
Chemical formula		He	He
Definition		Helium contains not less than 99.5 % V/V of helium. Applies to helium obtained by separation from natural gas supplies.	Helium contains not less than 99.0 % V/V of helium
Identification		Complies with the assay	The flame of a burning splinter of wood is extinguished. A small balloon filled with helium shows decided buoyancy
Tests intended for the manufacturer			
Assay	Specification	≥ 99.5 % V/V helium,	≥ 99.0 % V/V helium
	Analytical Method	Gas chromatography with thermal conductivity detector	Gas chromatography
Impurities			
CH ₄	Limit	≤ 50 ppm V/V	Not specified
	Analytical method	Infrared analyser	
O ₂	Limit	≤ 50 ppm V/V	Not specified
	Analytical method	Electrochemical cell	
H ₂ O	Limit	≤ 67 ppm V/V	Not specified
	Analytical method	Electrolytic hygrometer	
CO	Limit	Not specified	≤ 0.001 % V/V
	Analytical method		Detector tube
Air	Limit	Not specified	≤ 1.0 % V/V
	Analytical method		Determined in the assay
Odour	Limit	Not specified	No odour
	Analytical method		Organoleptic
Tests intended for the end user			
	Limit	Not specified	Not specified
	Analytical method		

5.11 Nitric oxide

Nitric oxide				
Monograph		Ph. Eur.	USP No equivalent US Pharmacopoeia monograph specified	
Name		Nitric oxide		
Reference		01/2008:1550		
Chemical Formula		NO		
Definition		Nitric oxide contains not less than 99.0% V/V of nitric oxide.		
Identification		Examine by infrared spectrometry and compare with the reference spectrum		
Tests intended for the manufacturer				
Assay	Specification	≥ 99.0 % V/V nitric oxide		
	Analytical method	Determine content of nitric oxide by difference using the mass balance equation after determining the sum of the impurities described under production.		
Impurities				
CO₂	Limit	≤ 3000 ppm V/V		
	Analytical Method	Gas chromatography with thermal conductivity detector		
N₂	Limit	≤ 3000 ppm V/V		
	Analytical method	Gas chromatography with thermal conductivity detector		
NO₂	Limit	≤ 400 ppm V/V		
	Analytical method	UV spectrophotometry analyser		
N₂O	Limit	≤ 3000 ppm V/V		
	Analytical method	Gas chromatography with thermal conductivity detector		
H₂O	Limit	≤ 100 ppm V/V		
	Analytical method	Electrolytic hygrometer		
Tests intended for the end user				
	Limit	Not specified		
	Analytical Method			

5.12 Argon

Argon			
Monograph	Ph. Eur.		USP No equivalent US Pharmacopoeia monograph specified
Name	Argon		
Reference	01/2023:2407		
Chemical Formula	Ar		
Definition	Gas obtained by cryogenic fractional distillation of ambient air. Argon contains not less than 99.995% v/v of argon calculated by deduction of the sum of impurities found when performing the test for impurities and water content.		
Identification	Gas chromatography with thermal conductivity detector; and Verify that the gas is not oxygen using a paramagnetic analyser.		
Tests intended for the manufacturer			
Assay	Specification	≥ 99.995 % V/V argon	
	Analytical method	Deduction of the sum of impurities	
Impurities			
O₂	Limit	≤ 5 ppm V/V	
	Analytical method	Gas chromatography with discharge ionisation detector	
N₂	Limit	≤ 5 ppm V/V	
	Analytical method	Gas chromatography with discharge ionisation detector	
CH₄	Limit	≤ 5 ppm V/V	
	Analytical method	Gas chromatography with discharge ionisation detector	
H₂O	Limit	≤ 10 ppm V/V	
	Analytical method	Electrolytic hygrometer	
Tests intended for the end user			
	Limit	Not specified	
	Analytical method		

5.13 Carbon monoxide

Carbon monoxide		
Monograph	Ph. Eur.	USP No equivalent US Pharmacopoeia monograph specified
Name	Carbon monoxide	
Reference	01/2011:2408 corrected 7.2	
Chemical Formula	CO	
Definition	Gas obtained by steam reforming (catalytic oxidation) of hydrocarbons. Carbon monoxide contains not less than 99.5% V/V of carbon monoxide.	
Identification	Infrared absorption spectrophotometry or it complies with the limits of the assay.	
Tests intended for the manufacturer		
Assay	Specification	≥ 99.5 % V/V carbon monoxide
	Analytical method	Infrared analyser
Impurities		
CO₂	Limit	≤ 300 ppm V/V
	Analytical method	Gas chromatography with thermal conductivity detector
CH₄	Limit	≤ 100 ppm V/V
	Analytical method	Gas chromatography with flame ionisation detector
H₂	Limit	≤ 300 ppm V/V
	Analytical method	Gas chromatography with thermal conductivity detector
Nickel tetracarbonyl / Iron pentacarbonyl	Limit	Not detectable
	Analytical method	Detector tube with limit detection of 0.1 ppm V/V
H₂O	Limit	≤ 10 ppm V/V
	Analytical Method	Electrolytic Hygrometer
Tests intended for the end user		
	Limit	Not specified
	Analytical method	

5.14 Carbon monoxide intermix (5 per cent) in nitrogen

Carbon monoxide intermix (5 per cent) in nitrogen			
Monograph		Ph. Eur.	USP
Name		Carbon monoxide intermix (5%) in nitrogen	No equivalent US Pharmacopoeia monograph specified
Reference		01/2018:2904	
Chemical formula		N/A	
Definition		Mixture containing 5% carbon monoxide (2408) in Low-oxygen nitrogen (1685)	
Identification		Carry out tests -A and C or - B and C A - Infrared absorption spectrophotometry B - Complies with limits of assay C - Gas chromatography with thermal conductivity detector	
Tests intended for the manufacturer			
Assay	Assay	95.0 per cent to 105.0 per cent of the nominal value of carbon monoxide (CO) in nitrogen (N ₂). NOTE This can be considered as containing between 4.75 % and 5.25% carbon monoxide	
	Analytical method	Infrared analyser	
Impurities			
H₂O	Limit	≤ 10 ppm V/V	
	Analytical method	Electrolytic hygrometer	
Tests intended for the end user			
	Limit	No tests section specified	
	Analytical method		

5.15 Methane

Methane				
Monograph		Ph. Eur.		USP No equivalent US Pharmacopoeia monograph specified
Name		Methane		
Reference		01/2015:2413		
Chemical Formula		CH ₄		
Definition		This monograph applies to methane obtained from natural gas and intended for medicinal use. Methane contains not less than 99.5% V/V of methane.		
Identification		Comparison of peak retention times of sample and reference gas obtained with gas chromatography		
Tests intended for the manufacturer				
Assay	Specification	≥ 99.5 % V/V methane		
	Analytical method	Gas chromatography with thermal conductivity detector		
Impurities				
N₂	Limit	≤ 500 ppm V/V		
	Analytical method	Gas chromatography with thermal conductivity detector		
C₂-C₄ Hydrocarbons	Limit	≤ 100 ppm V/V		
	Analytical method	Gas chromatography with flame ionisation detector		
H₂O	Limit	≤ 10 ppm V/V		
	Analytical Method	Electrolytic Hygrometer		
Tests intended for the end user				
	Limit	No tests section specified		
	Analytical method			

5.16 Methane intermix (2% per cent) in nitrogen

Methane intermix (2 per cent) in nitrogen			
Monograph		Ph. Eur.	USP No equivalent US Pharmacopoeia monograph specified
Name	Methane intermix (2%) in nitrogen		
Reference	01/2018:2905		
Chemical formula	N/A		
Definition	Mixture containing 2% methane (2413) in Low-oxygen nitrogen (1685)		
Identification	Carry out tests A or B A - Complies with limits of assay B - Comparison of peak retention times of sample and reference gas obtained with gas chromatography		
Tests intended for the manufacturer			
Assay	Assay	95.0 per cent to 105.0 per cent of the nominal value of methane (CH ₄) in nitrogen (N ₂). NOTE This can be considered as containing between 1.9 % and 2.1% methane	
	Analytical method	Gas chromatography with flame ionisation detector	
Impurities			
H₂O	Limit	≤ 10 ppm V/V	
	Analytical method	Electrolytic hygrometer	
Tests intended for the end user			
	Limit	No tests section specified	
	Analytical method		

5.17 Acetylene intermix (1 per cent) in nitrogen

Acetylene intermix (1 per cent) in nitrogen			
Monograph		Ph. Eur.	USP No equivalent US Pharmacopoeia monograph specified
Name	Acetylene intermix (1% in nitrogen)		
Reference	01/2018:2903		
Chemical formula	N/A		
Definition	<p>Mixture containing 1% Acetylene in Low-oxygen nitrogen (1685). The acetylene used in the manufacturing process is limited to acetylene produced by hydrolysis of calcium carbide. The method of storage of the acetylene is limited to cylinders filled with a porous mass and using acetone as a solvent. Prior to using the gas in the manufacturing process, the acetylene must be passed through an activated charcoal filter.</p>		
Identification	<p>Carry out tests -A or B A - Complies with limits of assay B - Comparison of peak retention times of sample and reference gas obtained with gas chromatography</p>		
Tests intended for the manufacturer			
Assay	Assay	95.0 per cent to 105.0 per cent of the nominal value of acetylene (C ₂ H ₂) in nitrogen (N ₂). NOTE This can be considered as containing between 0.95 and 1.05% acetylene	
	Analytical method	Gas chromatography with flame ionisation detector	
Impurities			
Acetone	Limit	≤ 250 ppm V/V	
	Analytical method	Gas chromatography with flame ionisation detector	
AsH₃	Limit	≤ 0.25 ppm V/V	
	Analytical method	Detector tube	
PH₃	Limit	≤ 0.2 ppm V/V	
	Analytical method	Detector tube	
H₂S	Limit	≤ 0.2 ppm V/V	
	Analytical method	Detector tube	
H₂O	Limit	≤ 10 ppm V/V	
	Analytical method	Electrolytic hygrometer	
Tests intended for the end user			
	Limit	No tests section specified	
	Analytical method		

6 Japanese Pharmacopoeia (18th Edition)

6.1 Oxygen

Oxygen		
Monograph	JP18	
Name	Oxygen	
Chemical Formula	O ₂	
Definition	Oxygen is oxygen produced by the air liquefaction separation method. It contains not less than 99.5 v/v% of oxygen.	
Description	Oxygen is a colourless gas under atmospheric pressure and is odourless.	
Identification	The retention time of principal peak obtained from oxygen is the same as that of the peak obtained from oxygen by gas chromatography.	
Purity		
N ₂	Limit	The peak area of nitrogen in the oxygen is not larger than that of the control sample.
	Analytical Method	Gas chromatography with thermal conductivity detector
Assay		
Assay	Specification	≥ 99.5% vol of O ₂ .
	Analytical method	Magnetic Analyser

6.2 Nitrous oxide

Nitrous oxide		
Monograph	JP18	
Name	Nitrous oxide	
Chemical Formula	N ₂ O	
Definition	Nitrous oxide contains not less than 97 vol% of nitrous oxide	
Description	Nitrous oxide is a colourless gas at room temperature and at atmospheric pressure and is odourless.	
Identification	1 A glowing splinter of wood held in nitrous oxide: it bursts into flame immediately. 2. The retention time of the main peak from nitrous oxide coincides with that of nitrous oxide by gas chromatography.	
Purity		
Acidity or alkalinity	Limit	Colour of the test solution is not deeper than the reference solutions
	Analytical method	Pass through acidified methyl red and bromothymol blue test solution in a Nessler tube. Compare colour against control solution
Reducing Substances	Limit	The colour is the same as the control solution
	Analytical method	Pass through potassium permanganate solution in a Nessler tube. Compare colour against control solution
Oxidising Substances	Limit	The colour is the same as the control solution
	Analytical method	Pass through potassium iodide-starch solution in a Nessler tube. Compare colour against control solution
Chloride	Limit	Turbidity produced does not exceed that produced in the control solution (can be calculated by the method)
	Analytical method	Pass through silver nitrate solution in a Nessler tube. Compare turbidity against control solution
CO₂	Limit	Turbidity produced does not exceed that produced in the control solution (can be calculated by the method)
	Analytical method	Pass through barium hydroxide in a Nessler tube. Compare turbidity against control solution of barium hydroxide containing sodium hydrogen carbonate
CO	Limit	No peak observed at the same retention time as that for carbon monoxide.
	Analytical method	Gas chromatography with thermal conductivity detector
Assay		
Assay	Specification	≥ 97.0 vol% of nitrous oxide
	Analytical Method	Gas chromatography with thermal conductivity detector

6.3 Carbon dioxide

Carbon dioxide		
Monograph	JP18	
Name	Carbon dioxide	
Chemical Formula	CO ₂	
Definition	Carbon dioxide contains not less than 99.5 vol% of carbon dioxide	
Description	Carbon dioxide is a colourless gas at room temperature and under atmospheric pressure. It is odourless.	
Identification	1 Put 100mL of carbon dioxide through a carbon dioxide measuring detector tube: the detector tube is changed to a stipulated colour tone by each detector tube, provided that the detector tube with an upper limit of measurement of not less than 10% is used. 2. Pass carbon dioxide into calcium hydroxide and a white precipitate is produced. Add acetic acid to the precipitate and it dissolves with effervescence.	
Purity		
Acidity	Limit	The test solution is not more coloured than the control solution.
	Analytical method	Pass through water in a Nessler tube and add methyl orange detector. Compare colour against control solution.
Reducing Substances*	Limit	The turbidity is the same as the control solution.
	Analytical method	Pass through silver nitrate solution in a Nessler tube. Compare turbidity against control solution.
CO	Limit	The concentration of carbon monoxide is less than 15 ppm according to each detector tube.
	Analytical method	Carbon monoxide measuring detector tube
Assay		
Assay	Specification	≥ 99.5 vol% of carbon dioxide.
	Analytical method	Volumetric gas absorption apparatus

* Reducing substances includes test for phosphine (PH₃) hydrogen sulphide (H₂S) and reducing organic substances.

6.4 Nitrogen

Nitrogen		
Monograph	JP18	
Name	Nitrogen	
Chemical Formula	N ₂	
Definition	Nitrogen is the nitrogen produced by the air liquefaction separation method. It contains not less than 99.5 vol% of nitrogen.	
Description	Nitrogen is a colourless gas at room temperature and under atmospheric pressure and is odourless.	
Identification	The principal peak obtained from nitrogen has the same retention time with the peak from nitrogen by gas chromatography.	
Purity		
O ₂	Limit	The peak area of oxygen obtained from nitrogen in the assay is not larger than 1/2 times that obtained from the standard gas mixture.
	Analytical method	Gas chromatography
Assay		
Assay	Specification	≥ 99.5 vol% of nitrogen.
	Analytical method	Gas chromatography with thermal conductivity detector